Acta Cryst. (1962). 15, 1306

A note on the elastic properties of pentaerythritol. By R. C. Srivastava*, Physics Department, University of Allahabad, Allahabad, India

(Received 10 April 1962)

Srivastava & Chakraborty (1960) published theoretical relationships connecting the rekha constants (using the nomenclature of Ramachandran & Wooster, 1951) corresponding to simple and elementary directions of reciprocal lattice and the thermal wave vector in terms of the elastic constants for crystals belonging to tetragonal crystal classes 4, $\bar{4}$ and 4/m. They also suggested a method of 'successive approximations' to obviate the difficulty arising in the evaluation of the elastic constants due to the presence of the constant C_{16} in the elastic matrix of the above crystals. It may be mentioned here that the elastic constants of any crystal belonging to 4, $\bar{4}$, or 4/m class have not yet been determined. It is expected, therefore, that the elastic constants of such crystals may prove to be of interest.

Pentaerythritol belongs to $\overline{4}$ crystal class. Its (200), (202) and (004) diffuse reflections with [010] axis and, (200) and (220) with [001] axis perpendicular to the direction of incidence of X-rays have been studied. The choice of the reflections was made according to the criterion given by Ramachandran & Wooster (1951). For evaluating the elastic constants from the experimentally determined rekha constants the method suggested by Srivastava & Chakraborty (1960) has been employed.

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The value of the constants is given below in units of 10^{10} dynes/cm.²:

$$C_{11} = 6 \cdot 1$$
, $C_{33} = 8 \cdot 0$, $C_{44} = 3 \cdot 5$, $C_{66} = 4 \cdot 6$, $C_{16} = -0 \cdot 39$, $C_{13} = 0 \cdot 50$, $C_{12} = -2 \cdot 50$.

The discussion of the accuracy of the method has been done in detail by Ramachandran & Wooster (1951) and Chakraborty & Sen (1958). The accuracy of the determination of the constants C_{11} , C_{33} , C_{44} , C_{66} , C_{16} is estimated to be $\pm 4\%$ whereas that of C_{12} and $C_{13} \pm 6\%$.

The author wishes to thank Prof. K. Banerji for his interest and encouragement and Dr S. C. Chakraborty for his constant help. The author also wishes to thank the Council of Scientific and Industrial Research for financial assistance.

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Acta Cryst. (1962). 15, 1306

The morphology and space groups of some halogen substituted derivatives of 5α-cholestane. By H. Geise, C. Romers and P. Hartman, Laboratory of Organic Chemistry; Geological and Mineralogical Institute; The University of Leiden, The Netherlands

(Received 18 July 1962)

In connection with investigations in one of our laboratories (L.O.C.), concerning the stereochemical features of 1,2 dihalogenocyclohexane and dioxane compounds (Kwestroo, Meyer & Havinga, 1954; Altona, Romers & Havinga, 1959; Van der Linden, 1958; Wessels, 1960) and of steroids (Van Moorselaar, 1962), a morphological survey and an X-ray analysis of some dihalogenosteroids was undertaken.

Single crystals of

I 2β , 3α [a, a] dibromo-cholestane

11 2β , 3α [a, a] chloro-bromo-cholestane

III 2β , 3α [a, a] dichloro-cholestane (two forms α and β)

IV 2α , 3β [e, e] dibromo-cholestane

V 2α , 3β [e, e] dichloro-cholestane

were obtained by slow crystallization from ethylacetate.

Morphology and optics

I. 2, 3 [a, a] Br, Br

$$a:b:c=4.214:1:0.5810; \beta=90^{\circ}; \text{ orthorhombic.}$$

Crystals are tablets {100} of varying thickness and with a rectangular outline. In the zone [010] the form {101} is always present and sometimes {001}. In [001] occur

 $\{110\}$, $\{210\}$ and $\{310\}$ of which the latter is always very small.

Barker angles, omitting {310}:

$$cr = 82^{\circ} 9', \ am = 30^{\circ} 9', \ bq = 13^{\circ} 21'.$$

Transformations: old \rightarrow new $001/010/\overline{1}00$; new \rightarrow old $00\overline{1}/010/100$.

Barker angles for combinations without {110}, but with {210} present:

$$cr = 16^{\circ} 12'$$
, $am = 25^{\circ} 23'$, $bq = 82^{\circ} 9'$.

Transformations: old \rightarrow new $020/\overline{1}00/001$; new \rightarrow old $0\overline{2}0/100/002$.

II. 2, 3 [a, a] Cl, Br

$$a:b:c=1.285:1:0.949$$
; $\beta=112^{\circ}\ 26'$; monoclinic.

Crystals are tablets $\{100\}$, elongated along the c-axis. Observed forms: $\{100\}$ large, $\{110\}$ and $\{1\overline{1}0\}$ small; sometimes $\{010\}$ and $\{0\overline{1}0\}$. The tablets are terminated by $\{001\}$, $\{101\}$, $\{0\overline{1}1\}$, $\{011\}$, $\{1\overline{1}1\}$ and $\{111\}$. Of these $\{0\overline{1}1\}$ is always larger than $\{011\}$, while $\{111\}$ always larger than $\{1\overline{1}1\}$. The latter six forms are not always all present, but various combinations occur.